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Development of the Acid and Base E-Module Based on Contextual Teaching And Learning for Class XI SMA/MA

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ABSTRACT

Acids and bases are one of the chemistry learning materials that are studied in class XI SMA/MA. Acids and bases are contextual and have a lot of relevance in everyday life. Additionally, acids and bases are prerequisites for studying some other chemistry. Therefore, appropriate learning models or strategies and teaching materials are needed. One of them is the use of contextual learning-based e-modules. This study aims to determine the validity, practicality, and on effectiveness of developing an acid-base e-module based on contextual teaching and learning for class XI SMA/MA. This form of study is an educational design research that uses the Plomp model which consists of three stages, namely the preliminary research, the prototyping phase, and the assessment phase. The assessment phase is not carried out. The results showed that e-modules had content, construct, and media validity values of 0.87, 0.88, and 0.89 respectively with valid categories, practicality according to teachers 88.52% with very practical categories and practicality according to students 85.10% with practical category. The average N-gain value for the effectiveness test is 0.7 with the high effectiveness category. The results of this study indicate that the developed e-module is valid, practical, and effective.

1. Introduction

The development of global technology has affected all aspects of human life, one of which is education (Ngafifi, 2014). Through technological advances in education, it is possible to change the orientation of learning from conventional learning to digital learning. Digital learning can be applied, one of which is in the presentation of learning content, such as e-module-based digital learning content (Kurniawan & Kuswandi, 2021).

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E-modules are independent teaching materials that are systematically arranged in certain learning units and presented in an electronic format. In learning activities, e-modules are connected by links as navigation that is more interactive in adding to the student's learning experience, and is equipped with video, animation, image, or audio presentations so that students can become more active in learning (Kemendikbud, 2017). The use of technology in learning can improve learning outcomes, teaching effectiveness, and can influence what and how learning should be studied and taught (Nugrahaeni et al., 2017; Nurkolis & Muhdi, 2020). The use of e-modules in learning is also in accordance with the demands of the 2013 curriculum development prevailing in Indonesia, where e-modules can change student learning orientation from teacher-centered learning to student-centered learning (Najuah et al., 2020).

One of the learning materials that can use e-modules in learning is chemistry. Chemistry is a science that studies the properties of matter and the changes it undergoes (Chang, 2010). One of the chemicals for class XI SMA/MA is acids and bases. Acids and bases are prerequisite materials for several chemistry materials in SMA/MA such as acid-base titration, buffer solutions and salt hydrolysis. If students do not understand the acid-base material well, then students will have difficulty understanding the next material. Acids and bases are contextual and have a lot of relevance in everyday life. Acid and base compounds are found in many household products, food, medicines, and of course almost every chemical laboratory (Tro, 2011). Based on this, it is appropriate for students to understand and be able to relate the acid and base material they are learning to everyday life. To be able to apply acid and base material in everyday life requires a strategy or learning model. One of the learning models that connects learning materials with real-life situations is Contextual Teaching and Learning (Antara et al., 2019; Zulaiha, 2016).

Contextual teaching and learning (CTL) is a learning strategy that fully involves students to be able to find the material being studied and be able to connect and apply it in real life (Sanjaya, 2010). With students being able to connect the material they learn with real life, students will be able to find the meaning or purpose of what they are learning, thus making students more interested in learning (Johnson, 2002). Contextual learning is different from traditional learning. Where in traditional learning, the main role of the teacher is to convey facts and procedures. The role of students is to memorize facts and practice the procedure. In contrast, in contextual learning the teacher's role is expanded, namely the teacher creates various learning experiences with a focus on understanding rather than memorizing (Crawford, 2001).

The results of Nuratika's research, et al (2020), reveal that the use of the CTL can improve student learning outcomes and students become more enthusiastic and active in learning, because students can more easily understand the subject matter with the linkage of the material they learn with real life. The results of Roziyah and Haryani's research (2017) also reveal that learning with the CTL. The model can increase student motivation and learning outcomes. Research by Dewi and Primayana (2019) revealed that there are significant differences in conceptual

understanding between students equipped with contextual-based learning modules and ordinary learning modules. Modules with CTL learning can improve students' conceptual understanding, because the CTL-based learning model emphasizes the process of constructing knowledge. In other words, the contextual learning model emphasizes more on providing direct experience to students to get to know more about nature so that students are more interested in the learning process. In this case, it is important for educators to strengthen student involvement in the material being taught by helping students associate the relationship between the knowledge they already have and their application in everyday life.

Based on the results of the questionnaire given to five SMA/MA chemistry teachers, namely MAN 2 Kerinci, SMAN 1 Kerinci, SMAN 7 Padang, SMAN 8 Padang, and SMAN 12 Padang, it was found that the teaching materials used by teachers in schools were generally still in the form of printed teaching materials, and some also use video and PPT. However, some students do not have these teaching materials. The teaching materials used by the teacher are also not based on contextual learning, only in the form of teaching materials commonly used in general. The teaching materials used by the teacher have also not been able to overcome the difficulties of some students in understanding the concepts being taught. The teacher also revealed that all schools were equipped with adequate ICT facilities and infrastructure, and the teacher also agreed to develop e-module teaching materials based on contextual learning on acids and bases.

Meanwhile, based on the results of questionnaire analysis from 60 students at SMAN 7 Padang, SMAN 8 Padang, SMAN 12 Padang, and MAN 2 Kerinci, it was obtained data that some students, namely 50% of students still considered acid and base material difficult, 95% of students wanted material more interesting teaching materials, and 93% of students agree with the use of teaching materials in the form of e-modules in chemistry learning, especially on acids and bases. Based on the data and problems above, the purpose of this research is to develop an acid and base e-module based on contextual teaching and learning for class XI SMA/MA which is valid, practical, and effective.

2. Methodology

This type of research is educational design research. This research will produce a product in the form of an acid and base e-module based on contextual teaching and learning for class XI SMA/MA and an analysis of the validity, practicality, and effectiveness of the e-module. The subjects in this study were three FMIPA UNP's chemistry lecturers, two SMA/MA's chemistry teachers at MAN 2 Kerinci and SMAN 1 Kerinci, and 16 students in class XI SMA/MA at MAN 2 Kerinci. Meanwhile, the object in this study is the E-module based on Contextual Teaching and Learning. The research model used is the Plomp development model (2013) developed by Tjeerd Plomp, which consists of three stages, namely the preliminary research stage, the prototyping phase, and the assessment phase. However, due to time and cost constraints, this research was limited to the prototyping phase. At the preliminary research stage, needs and context analysis,

literature study and conceptual framework development were carried out. At the prototyping phase, a prototype e-module design based on Contextual Teaching and Learning on acid and base materials for class XI SMA/MA. For each prototype, a formative evaluation will be carried out to improve and refine the developed e-module. This prototype building stage produces four prototypes. Figure 1 shows the overall design and research procedures of the Plomp model in this study.

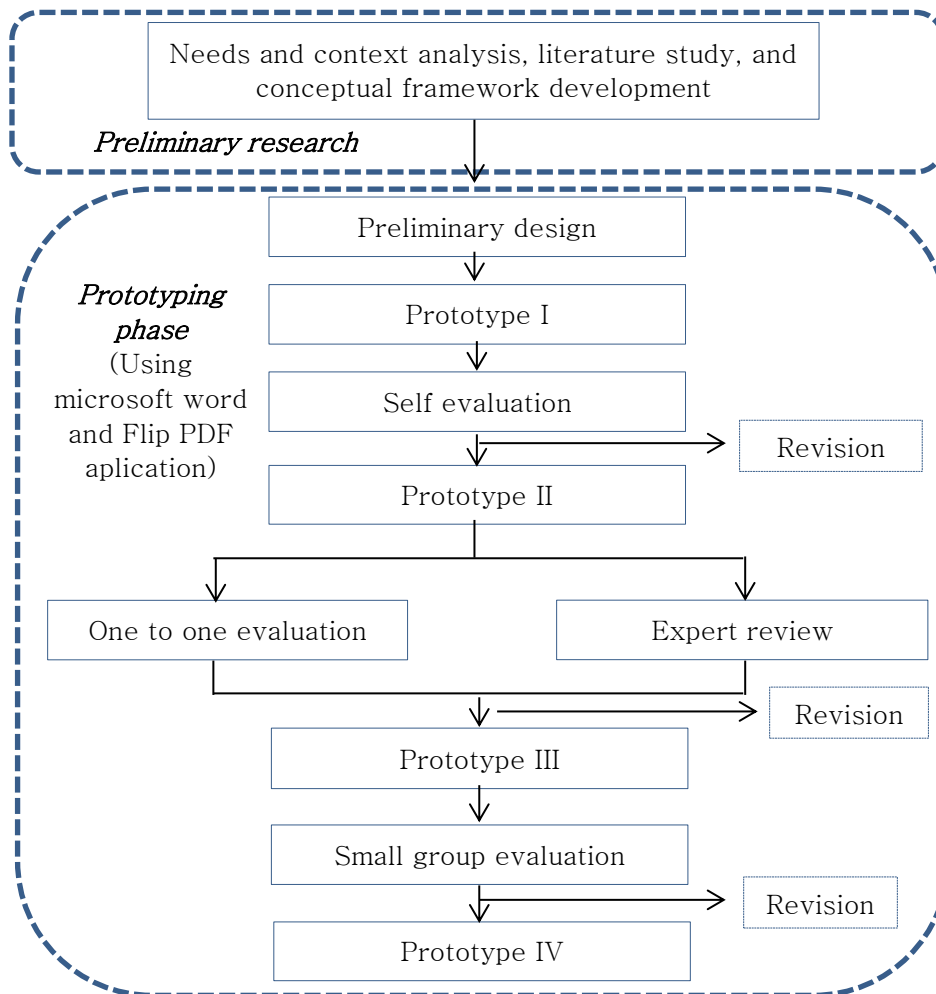


Figure 1. Plomp Development Procedure Modified from Plomp (2013)

The research data were analyzed using descriptive statistical techniques to obtain the average and percentage figures.

1. Validity

In the validity test, an instrument in the form of a questionnaire was used which was analyzed by the Aiken's V formula. Table 1 shows the validity assessment criteria based on the Aiken's V scale.

$$V = \frac{\sum s}{n(c-1)}$$

$$s = r - lo$$

Description:

V = Coefficient of validity

lo = The lowest value of the validity assessment ($lo = 1$)

c = The highest score of validity assessment ($c = 5$)

r = Validator category score

n = Number of validators

Table 1. Right-Tail Probabilities (p) for selected Values of the Validity Coefficient (V) (Aiken, 1985)

No. of Items (m) or Raters (n)	Number of Rating Categories (c)											
	2		3		4		5		6		7	
	V	p	V	p	V	p	V	p	V	p	V	P
2							1.00	.040	1.00	.028	1.00	.020
3							1.00	.008	1.00	.005	1.00	.003
3			1.00	.037	1.00	.016	.92	.032	.87	.046	.89	.029
4					1.00	.004	.94	.008	.95	.004	.92	.006
4			1.00	.012	.92	.020	.88	.024	.85	.027	.83	.029
5			1.00	.004	.93	.006	.90	.007	.88	.007	.87	.007
5	1.00	.031	.90	.025	.87	.021	.80	.040	.80	.032	.77	.047

2. Practicality

In the practicality test, an instrument in the form of a questionnaire was used which was analyzed with the practicality percentage formula from Purwanto (2009).

$$\text{Practicality level percentage} = \frac{\text{score obtained}}{\text{max score}} \times 100\%$$

Practicality assessment criteria adapted from Purwanto (2009), as follows:

86% - 100% = Very practical

76% - 85% = Practical

60% - 75% = Quite practical

55% - 59% = Less practical

≤ 54% = Not practical

3. Effectiveness

The effectiveness test used instruments in the form of pre-test and post-test questions which were analyzed using the N-Gain formula. The steps of the N-Gain test are:

- 1) Calculated the score of each test taker based on the pretest score and posttest score.

$$\text{Student value} = \frac{\sum \text{the answer score obtained}}{\text{max score}} \times 100\%$$

- 2) The N-Gain value of each student uses the N-Gain formula equation.

$$N\text{-Gain} = \frac{\text{Posttest score} - \text{Pretest score}}{100 - \text{Pretest score}}$$

- 3) Calculate the average N-Gain of each student.

$$\text{Average N-Gain} = \frac{\sum N\text{-Gain of students}}{\text{number of students}}$$

The results of the average N-Gain calculation are then interpreted using the N-Gain criteria. Table 2 shows the N-Gain criteria of Hake (1998).

Table 2. N-Gain Criteria of Hake (1998)

N-Gain Criteria	Interpretation
$g \geq 0,7$	Hight
$0,7 > g \geq 0,3$	Medium
$g < 0,3$	Low

3. Results and Discussion

Preliminary research

The needs and context analysis stage aims to collect information about the problems faced related to teaching materials used in schools as well as to develop learning objectives, materials, and learning strategies, especially on acid and base materials so that it is necessary to develop teaching materials in the form of e-modules based on contextual teaching and learning. At the needs analysis stage, it is done by distributing questionnaires to five chemistry teachers and 60 students. This questionnaire distribution activity was carried out at MAN 2 Kerinci, SMAN 1 Kerinci, SMAN 7 Padang, SMAN 8 Padang, and SMAN 12 Padang. Based on the results of the questionnaire analysis, information was obtained that: (1) the teaching materials used by teachers in schools are generally still in the form of printed teaching materials, while learning has now entered the digital era. (2) some students do not have these teaching materials, (3) some students still find acid and base material difficult, so students still have difficulty understanding the concepts that exist in acid and base material, and (4) the unavailability of the acid and base e-module based on contextual teaching and learning in schools. In the context analysis, it is obtained by analyzing the abilities that must be mastered by students in the learning process in accordance with the 2013 revised 2018 curriculum. Table 3 shows the basic competencies of acid and base materials, as well as learning indicators that are derived based on these basic competencies.

Table 3. Basic Competencies and Indicators of Achievement in Acid and Base Material Competencies.

The basic competencies of acid and base materials are as follows.	
3.10	Explain the concepts of acids and bases and their strengths and ionizing equilibrium in solution.
4.10	Analyzing the trajectory of changes in the pH of several indicators extracted from natural materials through experiments.
Learning indicators that can be derived based on basic competencies are as follows.	
3.10.1	Explain the concepts of acids and bases based on the Arrhenius theory.

-
- 3.10.2 Explain the concepts of acids and bases based on the Bronsted-Lowry theory.
 - 3.10.3 Explain the concepts of acids and bases based on the Lewis theory.
 - 3.10.4 Describe the degree of acidity (pH) based on its ionizing equilibrium in solution.
 - 3.10.5 Explain the strength of acids and bases and their relationship to the degree of ionization and their ionization constants.
 - 3.10.6 Calculating the pH of a strong and weak acid/base solution.
 - 4.10.1 Analyzing the properties of acidic, basic, or neutral solutions based on the color changes of several indicators extracted from natural materials and litmus paper indicators.
 - 4.10.2 Predict the pH of a solution using chemical indicators.
 - 4.10.3 Measuring the pH of various solutions of weak acids, strong acids, weak bases, and strong bases with the same concentration using universal indicators of pH meters
-

The next step in this preliminary research stage is literature study that aims to collect useful information to support the research process. The results obtained information that the e-module was compiled based on the books of the Kemendikbud (2017), Depdiknas (2008), and Suryosubroto (1983), with the source material in the e-module being sourced from a textbook chemistry. The e-module was developed using a contextual teaching and learning strategy (Crawford, 2001). As for the Plomp development model, it is based on the book "An Introduction to Educational Design Research" (Plomp, 2013).

The next stage is the development of a conceptual framework as a guide in the next development step, namely the formation of a prototype. This is done to find out what are the main concepts that need to be studied in acid and base materials, such as: determination of acid and base properties of solutions, theory of acids and bases, degree of acidity (pH), strength of acids and bases, pH measurement of solutions, and calculation of the pH of an acid and a base solution.

Prototyping phase

Prototype I, the design and preparation of the CTL-based acid and base e-module was carried out based on the results of the literature study that had been carried out. The e-module is organized into six meetings, where each meeting contains five CTL learning strategies, namely relating, experiencing, applying, cooperating, and transferring. The five strategies are contained in the form of activity sheets and worksheets which are one of the components of the e-module. This stage produces Prototype I.

In the relating section, it is presented by giving an image or phenomenon that is known by students in general and is associated with acid and base material. In the experiencing section, activity sheets are presented in the form of activities that must be carried out by students, such as observing practical videos, observing illustration images, simulations, virtual lab activities, and problem solving activities. This activity is carried out so that students can build new knowledge through direct experience that is arranged in the learning process. In the applying, a worksheet containing questions is presented to apply an information or concept that has been obtained in the previous activity. The cooperating section is presented simultaneously with the experiencing section and the applying section. In this section, students work on activity sheets and worksheets in groups. In the

transferring section, questions are presented in the form of a new context or as enrichment so that students can learn to transfer the knowledge they have learned. Figure 2 shows an example of an e-module display in each section.

Relating (menghubungkan)

Perhatikan gambar berikut ini!



Gambar 17. (a) Kopi hitam; (b) Jus lemon
(Sumber: <https://id.123rf.com> dan <https://id.123rf.com>.)

Gambar di atas menunjukkan dua macam minuman yang sering kita nikmati dalam kehidupan sehari-hari, yaitu kopi hitam dan jus lemon. Tahukah kamu, bahwa kedua minuman tersebut sama-sama bersifat asam? Namun, yang harus diingat bahwa, walaupun sama-sama bersifat asam, tetapi tingkat keasamannya berbeda, bukan? Bagaimanakah kita menyatakan tingkat keasaman suatu zat?

Sorensen (1868-1939), seorang ahli kimia dari Denmark, mempunyai ide cemerlang tentang hal tersebut. Ia mengajukan konsep pH untuk menyatakan tingkat keasaman dari suatu larutan. Nah, tingkat keasaman ini ada hubungannya dengan kesetimbangan pengionannya dalam larutan.

Experiencing & Cooperating (mengalami & kerja sama)

Perhatikanlah gambar berikut ini secara bersama dengan anggota kelompokmu!

Asam Asam



Gambar 21. Daya hantar listrik larutan HCl
(Sumber: Nivaldo, 2011: 499)

Asam Lemah



Gambar 22. Daya hantar listrik larutan HF
(Sumber: Nivaldo, 2011: 500)

Applying & Cooperating (menerapkan & kerja sama)

Berdasarkan model diatas, diskusikanlah bersama temanmu untuk menjawab pertanyaan-pertanyaan berikut ini!

Model 1. Tetapan kesetimbangan air (K_w).

- K_w merupakan tetapan kesetimbangan air, pada suhu 25 °C nilai $K_w = 1 \times 10^{-14}$. Dalam air murni H_2O merupakan satu-satunya sumber ion, sehingga $[H^+] = [OH^-]$. Tentukanlah berapa konsentrasi dari ion H^+ dan OH^- dalam air pada suhu 25 °C!
- Berdasarkan jawaban dari soal nomor 1, maka suatu larutan dapat dikatakan bersifat netral jika: $[H^+] \dots [OH^-]$ (isi dengan tanda > atau < atau =)
- Jika kedalam air ditambahkan suatu asam, maka asam akan memberikan ion ke dalam larutan tersebut, sehingga akan meningkatkan $[H^+]$ di dalamnya. Akan tetapi tetapan ionisasi air (K_w) tetap berlaku. Jika $[H^+]$ dalam larutan tersebut = 10^{-2} M (pada 25°C), hitunglah konsentrasi $[OH^-]$ dengan menggunakan rumus tetapan kesetimbangan air (K_w)!
- Berdasarkan jawaban soal nomor 3, maka suatu larutan dikatakan asam jika: $[H^+] \dots [OH^-]$ (isi dengan tanda > atau < atau =)

Transferring (memindahkan)



Gambar 16. Terumbu Karang. (Sumber: <http://www.geogebra.org/m>)

Perhatikan ananda menyelam di laut? Nah, jika ananda pernah menyelam dilaut maka ananda akan melihat adanya terumbu karang. Terumbu karang merupakan salah satu potensi sumber daya laut yang ada di Indonesia. Sebagai negara kepulauan, Indonesia merupakan negara yang memiliki terumbu karang terbesar di dunia. Luas terumbu karang Indonesia mencapai 284,3 ribu kilometer persegi atau setara dengan 18 persen dari terumbu karang yang ada di seluruh dunia.

Pembentukan terumbu karang didasar laut melibatkan reaksi asam dan basa Lewis antara ion kalsium yang melimpah dilaut dan gas karbon dioksida yang mengionkan garam kalsium karbonat. Berdasarkan keterangan diatas jawablah pertanyaan-pertanyaan berikut.

- Apakah yang dimaksud dengan terumbu karang?
- Jelaskanlah dan tuliskanlah reaksi kimia terbentuknya terumbu karang didasar laut?
- Berdasarkan reaksi terbentuknya terumbu karang didasar laut, tentukan manakah yang bertindak sebagai asam Lewis dan basa Lewis!

Klik untuk Menjawab

Figure 2. Example of e-module display in each section.

Prototype II, a formative evaluation was carried out in the form of self-evaluation of the e-module design on prototype I which aims to see the completeness of the e-module components developed. From the results of self-evaluation with a checklist for the components of the e-module, it was found that this prototype does not require revision because the e-module components are complete. At this stage produce a prototype II.

Prototype III, a formative evaluation is carried out in the form of expert review and one-to-one evaluation. The expert assessment was carried out by five validators who are experts in chemistry to analyze the validity of the content, construct, and media of the developed e-module. Table 4 shows information on the content validity data of the e-module. The content of the CTL-based e-module on acid and base material is declared valid by the validator with an average V value of 0.87. This shows that the content of the e-module developed is in accordance with the demands of Basic Competence (BC) 3.10 and 4.10 acid and base material in the 2013 revised 2018 curriculum. A teaching material can be said to be valid if it is in accordance with the applicable curriculum (Hafsah et al., 2016; Wijayanti et al., 2016).

Table 4. Content Validity Result Data

Aspects assessed	V	Category
The contents of the e-module are in accordance with KD 3.10.	0,95	Valid
The contents of the e-module are in accordance with KD 4.10.	0,9	Valid
The competency achievement indicator to be achieved is in accordance with KD.	0,95	Valid
The learning objectives are in accordance with KD.	0,95	Valid
The material contained in the e-module is correct.	0,8	Valid
The E-module contains explanations, pictures, illustrations, which are suitable for acids and bases.	0,9	Valid
There is a suitability of the questions given with the material in the e-module.	0,8	Valid
The contents of the e-module are contextual/ show phenomena that are close to everyday life.	0,85	Valid
The contents of the e-module can add insight to students' knowledge.	0,9	Valid
The images provided can provide information on acid and base material to students.	0,9	Valid
The videos provided can provide information on acid and base material to students.	0,8	Valid
E-module can be used for student-centered learning.	0,8	Valid
E-modules can be used for contextual teaching and learning.	0,8	Valid
Average	0,87	Valid

The next validity test is construct validity. This construct validity questionnaire contains two components, namely the linguistic component and the presentation component. Table 5 shows information about the data from the construct validity results. The construct validity of the CTL-based acid and base e-module by the validator found the average Aikens V value of 0.88 with a valid category. This shows that the e-module is valid in terms of construct feasibility, namely in terms of language and presentation. From the assessment of the linguistic component with an average V value of 0.89, it shows that the e-module that has been developed is communicative, does not cause confusion, and uses good and correct Indonesian rules so that it is easy to understand. According to Handayani & Legi (2016), a teaching material must be prepared using clear sentences and language so that it is easily understood by its users. Meanwhile, from the assessment of the presentation components with an average V value of 0.87, it shows that the developed e-modules have been arranged systematically, coherently and clearly with e-module elements according to Suryosubroto (1983) and Kemendikbud (2017). In addition, e-modules have also been presented in accordance with 5 learning strategies of contextual teaching and learning that are systematically arranged Crawford (2001).

Table 5. Construct Validity Result Data

Aspects assessed	V	Category
Language Component	0,89	Valid
Presentation Component	0,87	Valid
Average	0,88	Valid

The next validity test is the validity of the media. This media validity questionnaire contains two aspects, namely graphic aspects and programming and utilization aspects. Table 6 shows information regarding the data from the validity of the media. The validity of the media from the CTL-based acid and base e-module by the validator found the average Aikens V value of 0.89 with a valid category. From the assessment of the graphic component with an average V value of 0.89, it shows that the type and size of the letters, layout, and images on the e-module can be observed clearly and the overall design and appearance of the e-module is attractive. Meanwhile, from the assessment of programming and utilization aspects with an average V value of 0.88, it shows that the e-module is appropriate in the programming aspect so that it is easy to use and can be utilized in chemistry learning. Daryanto (2013) a good e-module is user friendly.

Table 6. Media Validity Result Data

Aspects assessed	V	Category
Graphical component	0,89	Valid
Programming and utilization aspects	0,88	Valid
Average	0,89	Valid

Next is a one-to-one evaluation of three students. The results of the analysis from the one-to-one evaluation found that the e-module developed was good both in terms of design and appearance, as well as in terms of presentation of material, use of letters, and language, it was also clear coupled with the images, tables, videos and animations contained in the e-module. e-modules can help students understand the material. Based on the suggestions obtained from expert assessments and one-to-one evaluations, revisions were made to improve and produce valid products. This stage produces prototype III.

Prototype IV, evaluation of the valid prototype III is carried out. The evaluation was carried out through small group trials to determine the level of practicality and effectiveness. Assessment aspects of the practicality of e-modules consist of aspects of benefits, ease of use, and time efficiency. In terms of the practicality of the acid and base e-module by the teacher, the results obtained were 88.52% with a very practical category. Table 7 shows the results of processing the CTL-based acid and base e-module practicality data by the teacher. Meanwhile, in terms of the practicality of the acid and base e-module by students, the results obtained are 85.10% in the practical category. Table 8 shows the results of processing practical data on the CTL-based acid and base e-module by students. This shows that the e-module developed is easy to use, the language is easy to understand, the learning time is used more efficiently, and can help the teacher's role as a facilitator. This is in accordance with what was stated by Yuliani & Seragih (2017) that the teacher acts as a facilitator in learning activities. The use of teaching materials can assist teachers in delivering material, so that learning time becomes more efficient (Asmiyunda et al., 2018). In addition, the components of the e-module can help students understand the material and can increase students' interest and motivation in learning. E-modules can also help students understand the material (Groth et

al., 2018; Herawati & Muhtadi, 2018). A teaching material can be said to be practical if it has ease of use and is useful in the learning process (Sukardi, 2012).

Table 7. Results of Practicality Analysis by Teachers

Aspects assessed	P (%)	Category
Ease of use	85,55	Practical
Efficiency of learning time	90	Very practical
Benefit	90	Very practical
Average	88,52	Very practical

Table 8. Results of Practicality Analysis by Students

Aspects assessed	P (%)	Category
Ease of use	85	Practical
Efficiency of learning time	86,30	Very practical
Benefit	84	Practical
Average	85,10	Practical

Furthermore, the assessment of the effectiveness of the CTL-based acid and base e-module is seen based on the increase in students' understanding by comparing the results of the pre-test and post-test. The pre-test is a test conducted before students learn to use the acid and base e-module, while the post-test is carried out after learning using the acid and base e-module. The test results showed an increase in student scores before and after the use of e-modules. The average increase in the N-Gain students is 0.7 with the category of high e-module effectiveness. That is, the CTL-based acid and base e-module can improve students' understanding. Contextual learning can make students active in the learning process and able to improve student learning outcomes (Nanda et al., 2017; Susiloningsih, 2016). CTL-based e-modules can also improve the scientific literacy of high school students (Nurhasnah & Sari, 2020).

4. Conclusion

Based on the results of research and data processing that has been carried out, it can be concluded that the acid and base e-module based on Contextual Teaching and Learning for class XI SMA/MA which was developed through the Plomp model is valid, practical, and effective in improving student learning outcomes and can help students understand the concepts contained in acid and base chemistry.

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